

KINETIC RESOLUTION OF 4-HYDROXY-2-CYCLOPENTENONE BY RHODIUM-CATALYZED ASYMMETRIC ISOMERIZATION

M. Kitamura, K. Manabe,¹ and R. Noyori*

Department of Chemistry, Nagoya University, Chikusa, Nagoya 464, Japan

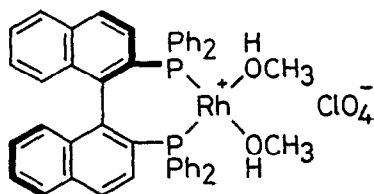
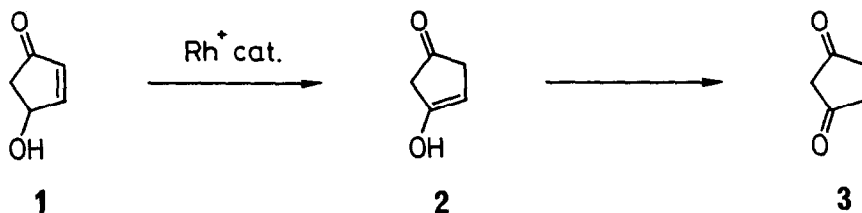
H. Takaya

Institute for Molecular Science, Myodaiji, Okazaki 444, Japan

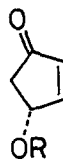
Summary: Cationic Rh—BINAP complexes catalyze isomerization of racemic 4-hydroxy-2-cyclopentenone to 1,3-cyclopentanedione with 5:1 enantiomeric discrimination.

(*R*)-4-Hydroxy-2-cyclopentenone [(*R*)-1] is a key chiral building block in our three-component coupling prostaglandin synthesis.² This compound has been obtained by the BINAL-H asymmetric reduction of 4-cyclopentene-1,3-dione,³ synthesis from (*2S,3S*)-tartaric acid,⁴ terrein,⁵ or 2,4,6-trichlorophenol (including resolution),⁶ chromatographic separation via the diastereomeric⁷ or racemic⁸ hydroxyl-protected derivatives, enzymatic kinetic resolution of the acetate⁹ or related compound,¹⁰ etc. We describe here the chemical kinetic resolution based on a new transition metal-catalyzed asymmetric 1,3-hydrogen migration.

In the presence of cationic Rh—phosphine complexes, the allylic alcohol **1** undergoes isomerization, leading to the keto enol **2** and ultimately the 1,3-diketone **3**. When racemic **1** was exposed to the (*R*)-BINAP complex, (*R*)-4,¹¹ the *S* enantiomer was consumed more readily. The asymmetric isomerization¹² in THF at 0 °C afforded $k_{\text{fast}}/k_{\text{slow}} = 5:1$ discrimination, providing a convenient method for resolution of readily accessible (\pm)-**1**¹³ to give the important *R* enantiomer. Thus the reaction with 0.5 mol% of (*R*)-**4** gave (*R*)-**1** in 91% ee at 72% conversion, and the extremely high crystallinity of the diketone **3** allowed easy



(*R*)-**4**



(*R*)-**1**, R = H

(*R*)-**5**, R = *t*-C₄H₉(CH₃)₂Si

separation of (*R*)-**1** from the reaction mixture. The enantiomerically pure enone was obtained by conversion to the crystalline *O*-silyl derivative, (*R*)-**5**.¹⁴ The 1,3-diketone **3** is also useful but difficult to make.¹⁵

The experimental procedure is as follows. A dry 300-mL Schlenk tube was charged with (*R*)-**4** (845 mg, 0.951 mmol) and anhydrous degassed THF (240 mL) under argon stream. To the resulting yellowish red solution cooled to -70 °C was added degassed (\pm)-**1** (16.9 g, 0.172 mol) by syringe, and the mixture was kept at 0 °C for 14 days. After separation of crystalline **3** (10.3 g), mp 151–153 °C, by filtration, the filtrate was subjected to column chromatography (silica gel, Fuji Devision Chemical Co., BW 300, 250 g; eluent, ether) to give (*R*)-**1** in 91% ee (4.60 g, 27% yield). This product (4.60 g, 46.9 mmol) was dissolved in dichloromethane (40 mL) and the solution was cooled to 0 °C. To this was added 4-dimethylaminopyridine (573 mg, 4.69 mmol), triethylamine (7.19 mL, 51.5 mmol), and *t*-butyldimethylsilyl chloride (7.77 g, 51.5 mmol), and the mixture was stirred at 26 °C for 2 h. After removal of the resulting precipitate by filtration, the filtrate was evaporated and the residue was diluted with hexane (300 mL). This was washed successively with water, saturated aqueous potassium hydrogensulfate solution, saturated aqueous sodium hydrogencarbonate, and brine. Drying over anhydrous magnesium sulfate, evaporation of the solvent, and distillation (80 °C/0.5 mmHg) afforded (*R*)-**5** (9.24 g, 93% yield). Recrystallization from pentane gave (*R*)-**5** in >99% ee, mp 29–30 °C (6.7 g, 18% yield from **1** (36% of theory)). HPLC analysis of **5**: column, DAICEL Chemical Industries Ltd. CHIRALCEL OC; eluent, 1% 2-propanol in hexane; flow rate, 0.5 mL/min; detection, 215-nm light; t_R 13.9 min (*R*), 17.8 min (*S*).

Acknowledgment. Valuable contribution of Mr. K. Nagai in searching the suitable reaction conditions is appreciated.

References and Notes

- Institute for Bio-Medical Research, Teijin Ltd.
- (a) Noyori, R.; Suzuki, M. *Angew. Chem., Int. Ed. Engl.* **1984**, *23*, 847. (b) Suzuki, M.; Yanagisawa, A.; Noyori, R. *J. Am. Chem. Soc.* **1985**, *107*, 3348.
- Noyori, R.; Tomino, T.; Yamada, M.; Nishizawa, M. *J. Am. Chem. Soc.* **1984**, *106*, 6717.
- Ogura, K.; Yamashita, M.; Tsuchihashi, G. *Tetrahedron Lett.* **1976**, 759.
- Mitscher, L. A.; Clark, G. W., III; Hudson, P. B. *Tetrahedron Lett.* **1978**, 2553.
- (a) Gill, M.; Rickards, R. W. *J. Chem. Soc., Chem. Commun.* **1979**, 121. (b) Gill, M.; Rickards, R. W. *Tetrahedron Lett.* **1979**, 1539.
- Suzuki, M.; Kawagishi, T.; Suzuki, T.; Noyori, R. *Tetrahedron Lett.* **1982**, *23*, 4057.
- Okamoto, Y.; Aburatani, R.; Kawashima, M.; Hatada, K.; Okamura, N. *Chem. Lett.* **1986**, 1767.
- Dohgane, I.; Yamachika, H.; Minai, M. *J. Synth. Org. Chem. Jpn.* **1983**, *41*, 896.
- For example, (a) Laumen, K.; Schneider, M. *Tetrahedron Lett.* **1984**, *25*, 5875. (b) Deardorff, D. R.; Matthews, A. J.; McMeekin, D. S.; Craney, C. L. *ibid.* **1986**, *27*, 1255. (c) Asami, M. *ibid.* **1985**, *26*, 5803.
- (a) Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R. *J. Am. Chem. Soc.* **1980**, *102*, 7932. (b) Miyashita, A.; Takaya, H.; Souchi, T.; Noyori, R. *Tetrahedron* **1984**, *40*, 1245. (c) Takaya, H.; Mashima, K.; Koyano, K.; Yagi, M.; Kumobayashi, H.; Taketomi, T.; Akutagawa, S.; Noyori, R. *J. Org. Chem.* **1986**, *51*, 629.
- For asymmetric isomerization of allylamines, see: Tani, K.; Yamagata, T.; Akutagawa, S.; Kumobayashi, H.; Taketomi, T.; Takaya, H.; Miyashita, A.; Noyori, R.; Otsuka, S. *J. Am. Chem. Soc.* **1984**, *106*, 5208. Review on asymmetric olefin isomerization: Otsuka, S.; Tani, K.; In *Asymmetric Synthesis*; Morrison, J. D., Ed.; Academic Press: New York, 1985; Vol. 5, Chapter 6.
- Tanaka, T. *Jpn. Kokai Tokkyo Koho* **1981**, 56-86128.
- Commercially available (Nissan Chemical Industries, Ltd., Tokyo, Japan) but expensive. The unit price depends on the purchase quantity.
- For synthetic utility, see: *Aldrichimica Acta* **1977**, *10*, 19. US\$16.30/g (Aldrich Catalog, 1986–1987). For other convenient synthesis, see: Suzuki, M.; Watanabe, A.; Noyori, R. *J. Am. Chem. Soc.* **1980**, *102*, 2095.

(Received in Japan 18 June 1987)